



Comment on “The origin of methane in the East Siberian Arctic Shelf unraveled with triple isotope analysis” by Sapart et al. (2017)

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Abstract. In this comment, we outline two major concerns regarding some of the key data presented in this paper. Both of these concerns are associated with the natural abundance radiocarbon-methane ($^{14}\text{C-CH}_4$) data. First, no systematic methodology is presented, nor previous peer-reviewed publication referenced, for how these samples were collected, prepared, and ultimately analyzed for $^{14}\text{C-CH}_4$. Not only are these procedural details missing, but the critical evaluation of them using gaseous and aqueous blanks and standards was omitted although these details are essential for any reader to evaluate the quality of data and subsequent interpretations. Second, due to the lack of methodological details, the source of the sporadic anthropogenic contamination cannot be determined and thus it is premature for the authors to suggest it was in the natural environment prior to sample collection. As the natural $^{14}\text{C-CH}_4$ data are necessary for the authors' stated scientific objectives of understanding the origin of methane in the East Siberian Arctic Shelf, our comment serves to highlight that the study's objectives have not been met.

1 Main text

In the article titled, “The origin of methane in the East Siberian Arctic Shelf unraveled with triple isotope analysis” (5 May, p. 2283, <https://doi.org/10.5194/bg-14-2283-2017>), Célia Sapart and coauthors present natural abundance radiocarbon-methane ($^{14}\text{C-CH}_4$) measurements from Laptev Sea sediments and waters alongside methane concentration and methane stable isotope measurements. The authors then draw conclusions about methane source–sink dynamics operating in this arctic shelf sea based upon these methane data.

Two concerns with the $^{14}\text{C-CH}_4$ data lead us to question whether these data should be used to interpret this natural system.

The first issue is that the method used to collect and prepare the $^{14}\text{C-CH}_4$ samples is inadequately described by Sapart et al. and there are no quality control data presented. Radiocarbon content of methane is not a routine measurement in natural waters because of the challenges associated with sampling and preparing a trace isotope of a trace gas. In the methods section of the article, the authors cite two techniques that relate only to the ^{14}C -accelerator mass spectrometry (AMS) analysis, while the methodologies used for the sample collection and preparation steps leading up to the ^{14}C analyses of sediment and seawater samples are absent. The natural $^{14}\text{C-CH}_4$ content of a sample can be affected by carbon and CH_4 added from the materials it encounters and by any contact with the atmosphere, so quality control measures are necessary to ensure that a sample is not significantly contaminated prior to analysis and that any minor contamination (i.e., blank addition) is accounted for in the final results. In the Supplement, the authors write that “none of the reference and blank measurements were abnormal,” without presenting any descriptions of or data stemming from these tests. Refereed techniques for collecting and preparing $^{14}\text{C-CH}_4$ samples from natural waters (Dean et al., 2017; Elder et al., 2018; Kessler and Reeburgh, 2005; Pack et al., 2015; Pohlman et al., 2000; Sparrow and Kessler, 2017) include detailed qualitative and quantitative descriptions of the measures taken to validate their methodologies. These measures include processing blank (methane-free) waters and treating methane-free gas and methane of known $^{14}\text{C-CH}_4$ content in the same way as samples. As the ^{14}C -AMS measurement error is typically very low relative to $^{14}\text{C-CH}_4$ collection and

preparation procedures, we can only assume that the error associated with the processes that most greatly affect the precision, sensitivity, and accuracy of the reported $^{14}\text{C-CH}_4$ signature is unaccounted for by the authors.

The second issue that calls the integrity of this study's $^{14}\text{C-CH}_4$ data into question is the existence of super-modern sediment and water column samples (approaching 100 times above modern) in the dataset. As the authors correctly reference, elevated $^{14}\text{C-CH}_4$ has previously been documented in other ocean waters (Kessler et al., 2008); however, the values presented here are up to 27 times higher than any previously reported elevated value. It is suggested in the main text and in the Supplement that the source of the "highly enriched ^{14}C " is anthropogenic and that it existed in the natural environment prior to sampling. We argue that it is premature to suggest an origin of this enriched ^{14}C , either environmental release or contamination (incurred during sample collection, processing, and/or analysis), when the $^{14}\text{C-CH}_4$ methodological details, with appropriate standards and blanks, are absent from the article. The possibility that the enriched ^{14}C was derived from the sampling equipment, vessel, and/or land-based laboratories was largely dismissed by the authors, while we attest that it is a valid option. The authors discount the possibility that their samples were contaminated during the sampling process "because no radioactive tracers were used during the sampling expeditions." This argument is untenable because the half-life of ^{14}C is 5730 years, meaning any surface contamination will persist for tens of thousands of years – well beyond the specific project in which it was used. In addition, the authors highlight that, for sediment samples, "the higher ^{14}C values correspond to the lower CH_4 concentrations" to suggest that a small amount of radioactive contamination in the environment was added to a variable background of naturally occurring CH_4 , which would most greatly affect the ^{14}C signature of the smallest sized (lowest CH_4 concentration) samples. This may be true, but another scenario that is also valid using the same logic is that the contamination was added during the $^{14}\text{C-CH}_4$ sample collection and/or preparation processes. This relationship was noted for sediment samples, but we are not informed in the article or Supplement on the relationship between CH_4 concentration and $^{14}\text{C-CH}_4$ content for the seawater samples. The lack of a data table containing the specific triple-isotope information for each CH_4 sample, in the article or in a data repository, has the effect of making this study unnecessarily opaque for a reader attempting to draw conclusions for themselves. The authors clearly state that additional experiments are necessary to determine the unknown origin of this isotopic enrichment; however, without those complimentary data, or at least data that prove it was in the sediments and waters prior to sample collection, its presence invalidates all $^{14}\text{C-CH}_4$ data presented in this study from contributing to our understanding of methane dynamics in the Arctic Ocean.

In a recently published study, we demonstrate how useful natural abundance $^{14}\text{C-CH}_4$ measurements can be towards

understanding the role of ancient sources of methane in arctic shelf seas (Sparrow et al., 2018). Importantly, in this study, we find that the stable isotope ($\delta^{13}\text{C-CH}_4$) and dissolved CH_4 concentration data, together, would suggest an entirely different (and, we argue, incorrect) interpretation of this system, which attests to the importance of $^{14}\text{C-CH}_4$ measurements for investigations into the origins of methane. When conducting natural abundance $^{14}\text{C-CH}_4$ studies, it is imperative that we do so using peer-reviewed published methods with appropriate radiocarbon blanks and standards; otherwise, interpretations made from $^{14}\text{C-CH}_4$ data are unverifiable and inconclusive.

Competing interests. The authors declare that they have no conflict of interest.

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