Silica cycling in the ultra-oligotrophic eastern Mediterranean Sea

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Abstract. Although silica is a key plant nutrient, there have been few studies aimed at understanding the Si cycle in the eastern Mediterranean Sea (EMS). Here we use a combination of new measurements and literature values to explain the silicic acid distribution across the basin and to calculate a silica budget to identify the key controlling processes. The surface water concentration of ∼1 µM, which is unchanged seasonally across the basin, was due to the inflow of western Mediterranean Sea (WMS) water at the Straits of Sicily. It does not change seasonally because there is only a sparse population of diatoms due to the low nutrient (N and P) supply to the photic zone in the EMS. The concentration of silicic acid in the deep water of the western Ionian Sea (6.3 µM) close to the S Adriatic are an of formation was due to the preformed silicic acid (3 µM) plus biogenic silica (BSi) from the dissolution of diatoms from the winter phytoplankton bloom (3.2 µM). The increase of 4.4 µM across the deep water of the EMS was due to silicic acid formed from in situ diagenetic weathering of aluminosilicate minerals fluxing out of the sediment. The major inputs to the EMS are silicic acid and BSi inflowing from the western Mediterranean (121 × 10^9 mol Si yr^-1 silicic acid and 16 × 10^9 mol Si yr^-1 BSi), silicic acid fluxing from the sediment (54 × 10^9 mol Si yr^-1) and riverine (27 × 10^9 mol Si yr^-1) and subterranean groundwater (9.7 × 10^9 mol Si yr^-1) inputs, with only a minor direct input from dissolution of dust in the water column (1 × 10^9 mol Si yr^-1). This budget shows the importance of rapidly dissolving BSi and in situ weathering of aluminosilicate minerals as sources of silica to balance the net export of silicic acid at the Straits of Sicily. Future measurements to improve the accuracy of this preliminary budget have been identified.

1 Introduction

Silicon is a key plant nutrient in the global ocean. There have been many studies describing the oceanic silica cycle (e.g. Benitez-Nelson et al., 2007; Brzezinski et al., 2011; DeMaster, 1981; Dugdale and Wilkerson, 1998; Nelson et al., 1995; Ragueneau et al., 2000). However, there have been few studies examining the unusual distribution of silica in the eastern Mediterranean Sea (EMS) (Crombet et al., 2011; Kress and Herut, 2001; Ribera d’Alcalà et al., 2003; Schink, 1967). The depth distribution of silicic acid (SA) in the global ocean is typically low in the photic zone with an increase in concentration with depth. This feature is generally interpreted as silicic acid uptake in the photic zone by diatoms (or other silica containing plankton) that sink where they dissolve and increase silicic acid concentration with depth (Armbrust, 2009). However there are problems with this simple explanation in the EMS because over large areas there are very low diatom numbers (Ignatiades et al., 2009; Psarra et al., 2000), assumed to be because the nutrient supply to the photic zone is below the threshold required for diatom growth. While Crombet et al. (2011) found evidence for what they describe as a “deep glass forest” of diatoms at the bottom of the deep chlorophyll maximum/upper nutricline across the Western Mediterranean Sea (WMS), they only found diatoms in the EMS at locations where there was a locally increased nutrient supply to the photic zone (Straits of Sicily, Mediterranean front and Cyprus eddy). Elsewhere, small non-silicate
phytoplankton, more efficient in utilising nutrients in limited supply, dominate (Huete-Ortega et al., 2011; Siokou-Frangou et al., 2010).

The concentration and seasonal cycle of dissolved silicic acid in the EMS is unusual compared to other areas of the global ocean. In the EMS silicic acid concentrations in the photic zone are generally relatively constant at \( \sim 1 \mu\text{mol Si kg}^{-1} \) and do not vary seasonally (Kress and Herut, 2001; Krom et al., 1992), which contrasts with the situation in many areas of the ocean, including the western Mediterranean (Brzezinski and Nelson, 1995; Marty et al., 2002). Silicic acid increases with depth with a sharp silicicline from 200 to \( \sim 1000 \) m. The maximum concentrations in the deep waters of the EMS are very low (6–11 \( \mu\text{mol Si kg}^{-1} \)) (Kress and Herut, 2001; Pujo-Pay et al., 2011), compared to 17–49 \( \mu\text{mol Si kg}^{-1} \) in the N Atlantic (Bainbridge, 1981), 80 \( \mu\text{mol Si kg}^{-1} \) in the S Atlantic and \( \sim 120 \mu\text{mol Si kg}^{-1} \) in the N Pacific (Chester, 2002).

There is a systematic increase in dissolved silicic acid across deep waters of the EMS from 6.3 \( \mu\text{mol Si kg}^{-1} \) in the western Ionian Sea close to the major source of Adriatic deep water (ADW) in the southern Adriatic to \( \sim 10.7 \mu\text{mol Si kg}^{-1} \) in the eastern Levantine basin in the direction of the deepwater circulation and increased water age (Roether and Schlitzer, 1991; Schlitzer et al., 1991). This eastwards increase is relatively much larger for silicic acid than that for nitrate or phosphate (Kress et al., 2003, 2011). Recently, as a result of the Eastern Mediterranean Transient (EMT) event, younger and denser water of Aegean origin with lower silicic acid content was introduced into the Ionian and Levantine deep layers (Roether et al., 1996, 2007). This reduced the concentration at depth and formed a pronounced mid-depth silicic acid maximum (Kress et al., 2003, 2011).

A key difference between the EMS and many other areas of the global ocean is that the annual phytoplankton bloom is in winter, at the same time as deep and intermediate waters are formed (Krom et al., 2013). Siokou-Frangou et al. (2010) describe increases in diatoms in February–March in the southern Adriatic, which is the source for deep water to the EMS (Adriatic deep water), and in the Cretan Sea, which is the source of Cretan Sea Outflow Water during the transient event. It is thus likely that when waters convect down from the photic zone, they contain diatom frustules made from opaline Si (BSi) as well as “preformed” silicic acid. In almost all areas of the global ocean, silicic acid is undersaturated and BSi tends to dissolve (DeMaster, 2004; Fanning and Schink, 1969). However this process is likely to be faster in the relatively “warm” and strongly undersaturated deep waters of the EMS.

Nutrient budgets have been used extensively to understand biogeochemical cycling processes in the eastern Mediterranean, partly because the system is well constrained with marine inputs and outputs occurring through the relatively narrow Straits of Sicily (Bethoux et al., 1992; Krom et al., 2004, 2010; Ribera d’Alcalà et al., 2003). The nutrient budget for N and P has been used to explain why the EMS has an unusually high nitrate : phosphate ratio (28 : 1) in deep water (Krom et al., 1991) and predict that there would be very limited N\(_2\) fixation (Krom et al., 2004), which was subsequently confirmed by field measurements (Ibello et al., 2010; Yoge et al., 2011). Only two silica budgets have been attempted in the Mediterranean. The first by Schink (1967) found a net loss of silicic acid at the Straits of Gibraltar and assumed this was balanced by riverine input of dissolved silicic acid, though noting that riverine flux data were very sparse at the time of his study. The second budget was carried out for the eastern Mediterranean by Ribera d’Alcalà et al. (2003), who found that the net flux of dissolved silicic acid out of the basin (150–250 \( \times 10^3 \) mol Si yr\(^{-1} \)) was far higher than the best estimates available for the inputs from rivers and other sources. They were unable to explain the reason for this pattern and hypothesised that the missing term in the silica budget might reflect some change in the terrestrial inputs and suggested that further work was needed to investigate this obvious mismatch. However these budgets were carried out only for dissolved silicic acid and did not include estimates for particulate biogenic opaline silica (BSi).

An addition and potentially important source of silicic acid to the EMS is the in situ chemical weathering aluminosilicate minerals (Fanning and Schink, 1969; Mackenzie and Garrels, 1965). This phase is recognised and measured as lithogenic Si (LSi) (Koning et al., 2002). This is likely to represent an important source of “new” silica to the water column in the EMS where there is a relatively large flux of Saharan dust which consists mainly of clay minerals and aluminosilicates (Venkataraman and Ryan, 1971) and the background concentration of silicic acid is equally low.

In this study we aim to explain the distribution and concentration of silicic acid in the eastern Mediterranean water column. Data from cross-basin water column transects are combined with previously unpublished porewater data to determine the importance of dissolved silicic acid fluxing from the sediment as a source to the deep water. A total silica budget for the EMS basin including both silicic acid and silica derived from particulate matter (BSi and chemical weathering of aluminosilicate minerals (LSi)) is calculated. Using this silica budget we identify the most important processes affecting the silica cycle and suggest where detailed measurements are required to improve the accuracy of the silica budget.

2 Materials and methods

2.1 Silicic acid in the water column

Data on the silicic acid distribution in the eastern Mediterranean were collected during several oceanographic cruises (Kress et al., 2003, 2014; Kress and Herut, 2001). Briefly, continuous profiles of pressure, temperature,
salinity, dissolved oxygen and fluorescence were performed with a Sea-Bird SBE 911 plus CTD system, interfaced to a SBE Carousel equipped with Niskin bottles. Water samples for silicic acid determination were collected in 15 mL acid-washed plastic scintillation vials and immediately frozen. In the laboratory, silicic acid was determined using a segmented flow Technicon AutoAnalyzer AA-II or Seal Analytical AA-3 using the reaction with ammonium molybdate in an acidic medium to form siliconolybdic acid which is reduced to the molybdenuim blue with stannous chloride, and the absorption is measured at 820 nm (Kress and Herut, 2001; Krom et al., 2003 using the reaction with ammonium molybdate in an acidic medium to form silicomolybdic acid which is reduced to the molybdenuim blue with stannous chloride, and the absorption is measured at 820 nm (Kress and Herut, 2001; Krom et al., 2003).

Average does not include core 10 because it was in shallower water (916 m), in the Aegean Sea, north of the island of Crete.

Table 1. Sampling stations, bottom silicic acid [SA] concentrations from hydrocasts in the water column of the deep EMS (Ionian and Levantine basins), and measurements from the upper layers of gravity cores used in calculations of diffusion gradients sampled during the TR-171 cruise of the R/V Trident in September 1975 (Jones, 1977).

<table>
<thead>
<tr>
<th>Stn no.</th>
<th>Location</th>
<th>Depth (m)</th>
<th>Core no.</th>
<th>Bottom [SA] (µM)</th>
<th>Upper slice thickness (cm)</th>
<th>Porewater [SA] in upper slice (µM)</th>
<th>Gradient (µM cm⁻¹)</th>
</tr>
</thead>
<tbody>
<tr>
<td>27</td>
<td>35°24.1’N</td>
<td>17°19.3’E</td>
<td>3800</td>
<td>Core 6</td>
<td>0.5</td>
<td>15.0</td>
<td>22.8</td>
</tr>
<tr>
<td>27</td>
<td>35°24.0’N</td>
<td>17°19.9’E</td>
<td>3800</td>
<td>Cast 6</td>
<td>9.3</td>
<td></td>
<td></td>
</tr>
<tr>
<td>29</td>
<td>34°16.9’N</td>
<td>19°32.0’E</td>
<td>3590</td>
<td>Cast 7</td>
<td>9.2&lt;sup&gt;a&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>30</td>
<td>34°25.5’N</td>
<td>20°07.9’E</td>
<td>2770</td>
<td>Core 7</td>
<td>0.5</td>
<td>20.0</td>
<td>43.2</td>
</tr>
<tr>
<td>34</td>
<td>34°00.3’N</td>
<td>23°10.9’E</td>
<td>2535</td>
<td>Cast 8</td>
<td>10.2</td>
<td></td>
<td></td>
</tr>
<tr>
<td>37</td>
<td>33°50.0’N</td>
<td>26°00.6’E</td>
<td>2720</td>
<td>Core 8</td>
<td>1.0</td>
<td>30.9</td>
<td>41.4</td>
</tr>
<tr>
<td>37</td>
<td>33°48.0’N</td>
<td>26°00.4’E</td>
<td>2755</td>
<td>Cast 9</td>
<td>1.0</td>
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<td>53.4</td>
</tr>
<tr>
<td>40</td>
<td>35°41.2’N</td>
<td>25°18.4’E</td>
<td>916</td>
<td>Core 10</td>
<td>8.5</td>
<td></td>
<td>35.8</td>
</tr>
</tbody>
</table>

<sup>a</sup> Bottom [Si] was assumed to be the same as at nearby station 29. <sup>b</sup> Average does not include core 10 because it was in shallower water (916 m), in the Aegean Sea, north of the island of Crete.

2.2 Silicic acid in the water column and porewaters sampled in 1975 (Jones, 1977)

The samples used for porewater flux determinations were obtained on cruise TR-171 of the R/V Trident during September 1975. The stations presented in this study were those stations in the Ionian and Levantine basins of the eastern Mediterranean (Table 1). Gravity cores were taken with a benthos gravity corer. The cores were capped and stored upright for at least 24 h in a cold room at in situ bottom temperatures before squeezing.

Inside the cold room, cores were sampled every 0.5 cm to a depth of 10 cm and then every 10 cm to the end. Each extruded sediment sample was placed in a Reeburgh type squeezer (Reeburgh, 1967) and the porewater allowed to drip directly into the storage bottle after 0.45 micron filtration. Details of the actual squeezer as used in four banks of five squeezer are given in Jones (1977) and Kruijvok and Manheim (1982). The porewater samples were sealed in clean polyethylene bottles with parafilm, and then capped and stored in the dark at room temperature for subsequent analysis. The sediment squeezer cake was sealed in a Whirl-Pak bag for subsequent porosity determination.

Hydrocasts were taken with 5 L Niskin bottles. The bottom sample at all stations was within 20 m of the sediment–water interface. Water samples were taken for dissolved oxygen, salinity and dissolved silicic acid determinations. Samples taken for salinity and oxygen determinations were processed on board ship. Both hydrocast and porewater samples were stored in clean polypropylene bottles in the dark and were analysed for silicic acid within 2 weeks of returning from the cruise. The automated method for the silicic acid determinations was a modification of the method of Truesdale and Smith (1976) using ascorbic acid instead of tin (II) chloride to reduce molybdoisilicic acid. Precision of three replicate hydrocast samples (n = 4) was 0.1 µM (1.2 %). The measured concentration in the water column during the TR-171 cruise (~9 µmol Si kg⁻¹) was similar in magnitude to subsequent water column measurements across the EMS.

Porosity was calculated from the calculated volume of the porewater/volume of the bulk sediment using a density for seawater of 1.028 and an average sediment density of 2.7 g cm⁻³.

Nomenclature: in this text we use “silica” (i.e. all silica phases) for the total budget, “biogenic silica” (BSi) for the particulate opaline phase and “silicic acid” for dissolved silicate. In addition we use units of µmol Si kg⁻¹ for the dissolved silicic acid phases except in calculations where we use for simplicity µmol Si L⁻¹. We assume the difference between these two units makes no significant difference to the various budget and flux calculations being carried out.
3 Results

3.1 Silicic acid in the water column across the basin

The surface waters of the offshore EMS generally have concentrations of \( \sim 1 \mu \text{mol Si kg}^{-1} \) silicic acid throughout the photic zone (Figs. 1–2) (Krasakopoulou et al., 1999; Kress and Herut, 2001; Lascaratos et al., 1999; Yilmaz and Tugrul, 1998). Until 1990, the concentration of SA increased with depth from the base of the photic zone to the bottom, except in the western Ionian Sea. There, a maximum silicic acid layer (8.3 \( \mu \text{mol Si kg}^{-1} \)) was found at 1200 m depth due to the intrusion of newly added ADW at depth toward the base of the western continental slope (Fig. 1). In 1987, the SA concentration in the deep waters of the western Ionian was 6.3 \( \mu \text{mol Si kg}^{-1} \), south of Crete 9.1 \( \mu \text{mol Si kg}^{-1} \) and in the eastern Levantine 10.7 \( \mu \text{mol Si kg}^{-1} \) (Fig. 2a). This represents an increase of 4.4 \( \mu \text{mol Si kg}^{-1} \) across the basin in the known direction of ADW flow (Roether and Schlitzer, 1991). The EMT event, which changed the deep circulation of the EMS, also altered the depth distribution of silicic acid. By 1999 its effect was already observed in the deep waters of the EMS as a layer of younger water of Aegean origin with lower silicic acid content (Fig. 2) (Kress et al., 2003). The concentration of silicic acid south of Crete was 7.6 \( \mu \text{mol Si kg}^{-1} \), 8.6 \( \mu \text{mol Si kg}^{-1} \) in the eastern Levantine and 8.3 \( \mu \text{mol Si kg}^{-1} \) in the western Ionian.

3.2 Silicic acid in porewaters

Figure 3 shows the concentration gradient of silicic acid in the porewaters as a function of depth for the upper 10 cm of all the cores taken by Jones (1977) within the EMS (Table 1). Stations 6, 7 and 8 have a relatively shallow and constant gradient which was rather different from the profile from station 10. These stations are from the deep EMS in areas where Crombet et al. (2011) found no evidence of increased diatom presence in the overlying water column. The upper porewater SA from cores 6, 7 and 8 have been used to calculate an average SA flux from the sediment in the basin to EMS deep waters. The SA concentration gradient driving this flux through the sediment–water interface was calculated using the interstitial [SA] from the uppermost slice of sediment on each core. A linear approximation was used. The deepest water column [SA] on a hydrocast at the same station as the core (or at a nearby station) was assumed to equal the [SA] at the sediment–water interface. To obtain the estimated gradient, that [SA] was then subtracted from the interstitial [SA] in the uppermost slice, and the difference divided by one-half of the thickness of the slice. The average gradient by this method was 35.8 \( \mu \text{M cm}^{-1} \). This value is probably a slight underestimate of the actual average gradient at the sediment–water interface. Interstitial [SA] profiles tend to show upward curvature near the sediment–water interface, suggesting that a non-linear approximation (e.g. Fanning and Pilson, 1974) might be used to obtain the gradient at the interface. The equation for an upward-curved [SA] distribution would yield a higher gradient at the interface than a linear approximation for the same distribution. Thus fluxes calculated with the average [SA] gradient for the EMS may be somewhat low.

As might be expected given the paucity of biogenic silica contributions to the EMS, the average value of the EMS [SA] gradient (35.8 \( \mu \text{M cm}^{-1} \)) was smaller than similarly obtained gradients from areas with established contributions of diatomaceous productivity: the Antarctic (up to 1450 \( \mu \text{M cm}^{-1} \); Fanning, unpublished data), the Gulf of Mexico (124 \( \mu \text{M cm}^{-1} \)) (Schink et al., 1974), the equatorial Pacific (45 \( \mu \text{M cm}^{-1} \) calculated from Hurd, 1973) or the Cariaco trench (52 \( \mu \text{M cm}^{-1} \)) (Fanning and Pilson, 1972). It is also somewhat lower than station 10, which is a location where diatoms have been found in the overlying water column (Ignatiades et al., 2009; Psarra et al., 2000).

The SA flux through the sediment–water interface was then calculated assuming only molecular diffusion. This was considered reasonable both because of the nature of the sediments, which are deep-sea sediments where bioirrigation was not expected to be a significant process, and because of the shape of the silicic acid profiles, which showed no evidence of being modified by depth-dependent bioirrigation. The molecular diffusion coefficient for silicic acid in seawater measured at 25 \( ^\circ \text{C} \) is \( 1 \times 10^{-5} \text{ cm}^2 \text{ s}^{-1} \) (Wollast and Garrels, 1971). Correcting this for temperature using the Stokes–Einstein relationship (\( T = 13 ^\circ \text{C} \) for ADW) = 0.7 \( \times 10^{-5} \text{ cm}^2 \text{ s}^{-1} \). This was then corrected for porosity and tortuosity using Eq. 1 (Berner, 1971):

\[
D_s = D_{SW}^3 \rho \theta^{-2}.
\]  

(1)

The average porosity for the upper centimetre of cores was 0.71, and using a tortuosity of 1.15 resulted in a calculated 
\[
D_s = 3.8 \times 10^{-6} \text{ cm}^2 \text{ s}^{-1}.
\]
The flux of silicic acid out of the sediment, $F_s$, calculated by Fick’s 1st law was

$$F_s = -\frac{dC}{dz} \times D_s.$$ 

(2)

Substituting in Eq. (2) gives

$$F_s = -(35.8 \mu M \text{ cm}^{-1}) \times (3.8 \times 10^{-6} \text{ cm}^2 \text{ s}^{-1})$$

$$= -4.3 \times 10^{10} \mu \text{ mol Si km}^{-2} \text{ yr}^{-1}.$$ 

This flux is similar to though somewhat lower than the fluxes measured in the S Adriatic ($6.6 \times 10^{10} \mu \text{ mol Si km}^{-2} \text{ yr}^{-1}$) from areas where diatoms are known to be present in the overlying waters (Giordani et al., 2002). This flux also similar to that calculated using core 10 ($6.4 \times 10^{10} \mu \text{ mol Si km}^{-2} \text{ yr}^{-1}$), which was the location where there were suggested to be diatoms in the overlying waters. Using an area of $1337 \times 10^9 \text{ m}^2$ (which is the area of the Ionian and Levantine basins together and is the X section shown in Fig. 2) the total silicic acid flux into the EMS was $57.2 \times 10^9 \text{ mol yr}^{-1}$.

4 Discussion

4.1 Distribution of silicic acid in the EMS prior to Eastern Mediterranean Transient

X-sectional profiles across the EMS show no change in SA content ($\sim 1 \mu \text{ mol Si kg}^{-1}$) in the photic zone (0–200 m; Fig. 2a and b) from the stations nearest to the Straits of Sicily, the source of inflowing surface water from the western Mediterranean (WMS) to the easternmost Levantine basin. Furthermore there is no evidence of seasonal changes in concentration (Kress and Herut, 2001). The inflowing water from the WMS has a concentration of $\sim 1 \mu \text{ mole Si kg}^{-1}$,
which is not altered by uptake due to diatoms in the surface waters because there are insignificant numbers of diatoms (Ignatiades et al., 2009; Psarra et al., 2000). In those locations where diatoms are known to exist, such as at the Mediterranean front (Crombet et al., 2011), it is possible to see decreases in SA due to diatom uptake.

The concentration of silica in the 200–500 m layer, which is the intermediate water exported from EMS, was 1–8 \( \mu \text{mol Si kg}^{-1} \) in summer (September–October) and somewhat higher in winter (1–10 \( \mu \text{mol Si kg}^{-1} \), Fig. 4, Table 2). This intermediate layer flows out of the EMS due to its unusual anti-estuarine circulation with a residence time of \( \sim 8 \) yr (Van Cappellen et al., 2014). It is the reason why the N and P concentrations are so low compared to other areas of the global ocean. It is likely that the SA content in the EMS is unusually low for the same reason. Prior to the formation of the EMT, there was a simple increase in silicic acid from west to east. Silicic acid increased from 6.3 \( \mu \text{mol Si kg}^{-1} \) in the western Ionian Sea to 10.7 \( \mu \text{mol Si kg}^{-1} \) in the eastern Levantine basin (Fig. 2), which also corresponds to the direction of water flow of ADW (Roether and Schlitzer, 1991). The "initial" silicic acid observed in the western Ionian Sea derives from deep-water formation in the southern Adriatic Sea. Unusually in the EMS, compared with most areas of the ocean, deep water is formed simultaneously with the annual phytoplankton bloom (Krom et al., 2013). The best estimate for dissolved silicic acid concentration in winter in the area of the S Adriatic where ADW forms is 3 \( \mu \text{mol Si kg}^{-1} \) (Zavatarelli et al., 1998). In addition there is biogenic particulate silica (BSi) derived from diatoms growing during the phytoplankton bloom in the S Adriatic in winter. Boldrin et al. (2002) measured the total suspended matter during the annual phytoplankton bloom as 0.88 mg L\(^{-1}\) in the S Adriatic close to the area of Adriatic deep-water formation. At a trap located at 150 m depth, they found that BSi represents 22\% of the total material flux. If we assume that the downward flux has the same proportion of BSi as the plankton on the water column, then the concentration in the water column is

\[
(0.88 \times 0.22) \times 1000/60, \quad \text{i.e.} \; 3.2 \mu \text{mol Si kg}^{-1}.
\]

Thus the initial silicic acid value of 6.3 \( \mu \text{mol Si kg}^{-1} \) in the western Ionian deep water is preformed silicic acid plus BSi present in the downwelled ADW, which dissolved rapidly. This is considered reasonable since the deep water in the EMS is not only low in silicic acid concentration compared to other deep water locations in the ocean but is also relatively warm (\( \sim 13^\circ \text{C} \)) (Roether et al., 2007). Though both factors increase the rate of BSi dissolution, temperature is considered most important (DeMaster, 2004).

As the deep water flows to the east, the total silicic acid increases in concentration by 4.4 \( \mu \text{mol Si kg}^{-1} \). This increase is considerably larger than the observed changes in nitrate in the deep water (\( \sim 0.5 \mu \text{mol N kg}^{-1} \)), which itself is larger.
than the observed change in phosphate (0.05 µmol P kg⁻¹; Krom et al., 2013). These differences correspond to known differences in recycling efficiency of nutrients in the water column in the EMS and elsewhere. In the EMS it has been shown that P is very efficiently recycled in the upper layers of the water probably because of the extreme P-starved nature of the system (Krom et al., 2005; Thingstad et al., 2005). Nitrogen is recycled somewhat less rapidly, but both are recycled faster than biogenic silica is converted into dissolved silicic acid (DeMaster, 2004).

In order to calculate the total amount of silicic acid supplied to the deep water, we use the definition of the eastern Mediterranean used by Van Cappellen et al. (2014) of the Ionian plus Levantine basin where there is a relatively simple flow of deep water from its source in the southern Adriatic to the far eastern Levantine basin (prior to the EMT). This part of the EMS has a total area of 1337 × 10⁹ m² and a total volume below 500 m of 17 × 10¹⁴ m³. Roether and Schlitzer (1991) suggest that the water column below 1200 m is well mixed, which is the approximate depth of the base of the silicicline. The total volume below 1200 m is 7.64 × 10¹⁴ m³. Thus the total increase in silicic acid below 1200 m is 7.6 × 10¹⁴ × 4.4 µmol Si L⁻¹ = 33.6 × 10¹¹ mol Si.

In addition the volume between 500 and 1200 m has increased silicic acid, which is supplied from below by eddy diffusion. Here we assume that the silicicline is linear and thus the average change in silicic acid is 2.25 µmol Si L⁻¹. The total amount of silicic acid supplied to the deep waters of the EMS was 33.6 × 10¹¹ mol Si + 9.36 × 10¹⁴ × 2.25 µmol Si L⁻¹ = 54.6 × 10¹¹ mol Si.

This change in silicic acid concentration takes place as the deep water flows from the southern Adriatic/western Ionian to the far SE Levantine basin. Roether and Schlitzer (1991) calculated the residence time of ADW to be 100–126 yr based on chlorofluorocarbon and tritium values. Using this calculated residence time of deep water in the EMS, the silicic acid content increases at a rate of

\[ \left( 54.6 \times 10^{11} \text{ mol Si in } 126-100 \text{ yr} = 43-54 \times 10^9 \text{ mol Si yr}^{-1} \right). \]

This is similar to, though slightly lower than, the calculated flux of silicic acid from the porewaters of the sediment underlying the EMS deep waters (57.2 × 10⁹ mol Si yr⁻¹). It is thus possible to explain the observed increase in dissolved silicic acid in the deep water to be mainly a result of dissolution of particulate silicic acid in the sediment and flux into the deep water (within ±20%). If there is silicic acid produced by dissolution in the water column as is found to be important in many areas of the world’s oceans (DeMaster, 2004), it is relatively small compared to the sediment source.

### 4.2 Silica budget calculations for the EMS

In this calculation, we use the best available estimates for the external dissolved Silicic acid into and out of the eastern Mediterranean basin for exchange through the Straits of Sicily, riverine and submarine groundwater input, atmospheric input and flux of silicic acid from porewaters (Table 4). We include BSi (and LSi) supplied externally to the basin as important parts of the biologically reactive silica budget that were not included in previous budgets (Ribera d’Alcâla et al., 2003; Schink, 1967). An indication that such an approach which includes BSi and LSi is justified is the calculation used to explain the observed increase in silicic acid in the deep water across the EMS which required both dissolved silicic acid and BSi in the descending ADW to be combined to create the observed initial silicic acid in the western Ionian sea.

#### 4.2.1 Fluxes through the Straits of Sicily

The silicic acid flux through the Straits of Sicily was calculated using water flux determined by Astraldi et al. (1999) together with the measured silicic acid concentration data from the Mediterranean Targeted Project II–Mass Transfer and Ecosystem Response (MTPII-MATER) data set (Lavezza et al., 2011). Although Astraldi et al. (1999) present the flux of water out of the eastern Mediterranean through the Tunisian and Sicilian passages separately, in this calculation only the summed monthly flux for both passages together was used. This was because the flux for each passage was similar (0.54 ± 0.58 and 0.55 ± 0.17 × 10⁶ m³ s⁻¹). In order to maintain the salinity balance of the basin, there must be 4% more water flowing into the eastern Mediterranean at the Straits of Sicily than flowing out (Bethoux, 1980). This value is similar to the estimate of total outflow through the Straits of Sicily of 1.2 × 10⁶ m³ s⁻¹ based on salinity balance. The calculated flux through the Straits of Sicily into the eastern Mediterranean was 1.13 × 10⁶ m³ s⁻¹ with an outflow of 1.08 × 10⁶ m³ s⁻¹. Astraldi et al. (1999) found that there are systematic seasonal changes in both water flux and the depth of the upper layer of the outflowing Levantine Intermediate Water (LIW) through the Straits of Sicily. The flux is higher in winter than in summer by approximately 50%. All the water which flows into the eastern basin is surface water with the nutrient characteristics of water within the photic zone. In winter (November–March) only intermediate and deep water below 200 m flows out of the eastern Mediterranean (Astraldi et al., 1999). For the rest of the year the outflowing water is much closer to the surface. Two calculations were carried out for this budget (Table 3). In the first, the total annual water flow was used: 1.13 Sv inflow and 1.08 Sv outflow. In the second calculation it was assumed that the monthly rate of flow was 1.5 times higher in winter than in summer. The MTP-MATER silicic acid data set—which was measured in September 1996 (MATER 1), October 1997 (MATER 3) and
November 1999 (MATER 7) – was used to calculate the concentration of silicic acid (and hence the annual flux) through the Straits of Sicily (Lavezza et al., 2011; Fig. 4). The data from September and October (Fig. 4a and b, Table 2) are similar to profiles measured elsewhere in the EMS in the summer (Kress and Herut, 2001), with a depth-averaged silicic acid concentration of 1.2 µmol Si kg\(^{-1}\) in the upper 100 m, which is the layer flowing into the EMS (Astraldi et al., 1999). Concentrations increased with depth to a relatively constant value of 6.2 µmol Si kg\(^{-1}\) (depth-averaged) similar to values found in the western Ionian sea (Figs. 2 and 4).

In addition to silicic acid fluxing through the Straits of Sicily there is also BSi. Crombet et al. (2011) found, in July 2009, 0.25 µmol BSi kg\(^{-1}\) in the upper (inflowing) water column with insignificant amounts in the deeper (outflowing) water. The Straits of Sicily are relatively shallow and are known to have considerable turbulence and eddies (Astraldi et al., 1999). It is thus likely to be a location of nutrient upwelling which elsewhere in the EMS is known to result in diatom growth (Crombet et al., 2011; Siokou-Frangou et al., 2010). It is however likely that the amount of BSi in the upper water column in July is a minimum for the year. In general, primary productivity in July in the EMS is close to its annual minimum (Krom et al., 2013). Here we assume that the annual average of BSi together with upwelled silicic acid not yet consumed by diatoms into BSi is 1.0 µmol Si kg\(^{-1}\). This represents a relatively high estimate for the BSi flowing into the EMS. Using this value, the net flux of silicic acid through the straits was 140 × 10\(^9\) mol Si yr\(^{-1}\) (Table 3).

However the silicic acid measured in November 1999 (Fig. 4c) showed considerable scatter in the upper 200 m of the water column, with an average value of 4.25 µmol Si kg\(^{-1}\). November is known, from remote sensing data and direct measurements, to be the beginning of deep winter mixing in the EMS and the start of the annual phytoplankton bloom (Krom et al., 2013; Patara et al., 2009). It is reasonable to suggest that this increased silicic acid was due to this deep winter mixing before most of the silicic acid was taken up by diatoms. No BSi was measured during the MATER cruises. Therefore in an alternative calculation we assume that the net silicic acid supplied to the upper 200 m in November was typical of the winter supply from below and allow this net concentration of silicic acid to be advected into the EMS with no BSi. In practice it does not matter whether this silicic acid was actually advected as dissolved silicic acid or was taken up as diatoms and advected as BSi. It is assumed that this winter flux occurred for 6 months and the summer flux with BSi of 1 µmol Si kg\(^{-1}\) occurred for 6 months. We also use the higher winter flow rate as determined by Astraldi et al. (1999), where the winter flow rate represents 60 % of the annual water flux and summer flow rate 40 %. Using these assumptions the net silicic acid from the basin was 127 × 10\(^9\) mol Si yr\(^{-1}\) (Table 3). We use this higher influx of silica in our total budget estimate (Table 4).

4.2.2 Sediment as a source and sink of silicic acid

The calculated flux of silicic acid from the offshore sediments into the water column was estimated as 57 × 10\(^9\) mol Si yr\(^{-1}\). We argue here that this silicic acid is formed dominantly by the diagenetic alteration of aluminosilicate minerals derived from dust, volcanic ash and other terrigenous sources (DeMaster, 2004; Fanning and Schink, 1969; Koning et al., 2002) and thus represents an additional external source. There are several active volcanoes in the EMS basin (e.g. Santorini), and it is known that there are observable ash layers found within the sediment. Even in the N Atlantic (the Iberian Shelf and the Angolan basin), where diatoms are abundant and the dissolved silicic acid in the deep water is much higher, an average of 50 % of the silicic acid formed in the porewaters and advected out is derived from the diagenesis of aluminosilicate minerals (Koning et al., 2002). If BSi were a major component in the particulate matter reaching the sediment in cores 6–8, that phase would be expected to dissolve more rapidly than lithogenically derived SA and would create a convex upward shape, possibly similar to the profile of core 10 (which is not included in the regional flux calculations). The upper porewater profiles for cores 6–8 are linear and have a low gradient, which is consistent with being derived from mineral weathering. Given that it is known that for large areas of the EMS there are relatively few diatoms (Ignatiades et al., 2009) and that the deep waters have very low SA concentration compared to most deep parts of the ocean, this assumption is considered reasonable. An alternative source for this SA might be a flux from the dissolution of BSi in the upper sapropel layer which is present at 20–30 cm depth (e.g. van Santvoort et al., 1996), below our porewater data. For the purposes of this budget this source would also be new silica into the basin. However no data are presently available on the long-term burial flux of biogenic or lithogenic Si in the eastern Mediterranean (Koning et al., 2002). We assume that no significant amounts of silica are permanently lost from the system by sedimentation. These assumptions clearly need to be tested by field measurements.

4.2.3 Riverine inputs

Ludwig et al. (2009) estimated the average flux of SA supplied to the EMS by riverine input between 1963 and 1998 as 23 × 10\(^9\) mol Si yr\(^{-1}\). Although there has been some variation in SA input over the time period considered (from 19 to 27 × 10\(^9\) mol Si yr\(^{-1}\) with a standard deviation of ~10 %), it is much less variable than inorganic N and P because, unlike them, SA is not considered a major pollutant. However in their calculation Ludwig et al. (2009) only included dissolved SA. Conley (1997) showed that measurement of silicic acid alone underestimates the global flux of silica from rivers to the ocean because it does not include BSi. Laruelle et al. (2009) estimate a total global flux of riverine dissolved silicic acid to be 6.2 Tmol Si yr\(^{-1}\) and a flux of
Table 2. Table showing the depth-averaged concentration of dissolved silicate (in µ mol kg\(^{-1}\)) measured for the series of stations across the Straits of Sicily sampled during the MATER-MTP cruises between 1996 and 1999 (data were obtained from Lavezza et al., 2011).

<table>
<thead>
<tr>
<th>Cruise name, time of sampling and number of stations averaged in brackets</th>
<th>Depth-averaged concentration of SA in layer flowing into the EMS; depth of layer in brackets</th>
<th>Depth averaged concentration of SA in the layer flowing out of the EMS</th>
</tr>
</thead>
<tbody>
<tr>
<td>MATER 1; Sep 1996 (24)</td>
<td>1.18 (0–100 m)</td>
<td>6.24 (&gt; 100 m)</td>
</tr>
<tr>
<td>MATER 3; Oct 1997 (17)</td>
<td>1.22 (0–100 m)</td>
<td>6.17 (&gt; 100 m)</td>
</tr>
<tr>
<td>MATER 7; Nov 1999 (12)</td>
<td>4.24 (0–100 m)</td>
<td>7.66 (&gt; 200 m)</td>
</tr>
</tbody>
</table>

Table 3. Table of calculated fluxes of silicic acid and BSi through the Straits of Sicily; see the text for detailed information of the assumptions made in each particular calculation. The values in bold were used in the total silica budget for the EMS (Table 4). All fluxes in units of \(10^9\) mol Si yr\(^{-1}\).

<table>
<thead>
<tr>
<th>Chemical species</th>
<th>Inflow</th>
<th>Net inflow</th>
<th>Outflow</th>
<th>Net flux</th>
</tr>
</thead>
<tbody>
<tr>
<td>Total annual flux assuming summer values for silicic acid and an average of 1 µmol BSi L(^{-1}) in the photic zone for the entire year.</td>
<td>47</td>
<td>39</td>
<td>85</td>
<td>225</td>
</tr>
<tr>
<td>Calculated summer flux (6 months) using summer values for silicic acid and 1 µmol BSi L(^{-1}) with 40% of the total annual flow (Astraldi et al., 1999).</td>
<td>19</td>
<td>16</td>
<td>34</td>
<td>90</td>
</tr>
<tr>
<td>Calculated winter (6 months) using winter (Nov 1999) silicic acid values and with 60% of the total annual flow (Astraldi et al., 1999).</td>
<td>103</td>
<td>103</td>
<td>174</td>
<td>71</td>
</tr>
<tr>
<td>Calculated total annual fluxes summing winter and summer values.</td>
<td>121</td>
<td>16</td>
<td>137</td>
<td>264</td>
</tr>
</tbody>
</table>

BSi to be 1.1 Tmol Si yr\(^{-1}\). In the absence of regional data (e.g., river Po) we use this global ratio to correct the riverine flux of \(23 \times 10^9\) mol Si yr\(^{-1}\) to a total riverine input of \(27 \times 10^9\) mol Si yr\(^{-1}\).

4.2.4 Submarine groundwater flux (SGW)

A potentially important source of nutrient input into the EMS is submarine groundwater discharge. Large areas of the coastal Mediterranean are made from chalk and other porous and/or permeable rocks. Laruelle et al. (2009) estimate that the submarine discharge of freshwater into the Mediterranean represents 25% of the total riverine input. It is known that SGW contains elevated amounts of dissolved nutrients, including silicic acid. They estimate a global average of 200 µmol Si kg\(^{-1}\), which is in the mid-range of a specific study on SGW carried out at Dor on the Israeli coast (Weinstein et al., 2011). Taken together this represents a flux of 9.7 mol Si yr\(^{-1}\) calculated as 25% of total riverine water flow \(\times 200\) µmol Si kg\(^{-1}\).

4.2.5 Atmospheric flux

Saharan dust is a major source of particulate matter to the offshore EMS. This dust falls through the water column relatively rapidly and becomes a major component of aluminosilicates in the sediment which undergo chemical weathering. Thus it is only the short-term (water column) dissolution of silicic acid which is relevant to this budget calculation.

In a short-term dissolution experiment (48 h) dust collected from Crete (and used in a MESOQAUA dust addition experiment) released 25 nmol Si mg\(^{-1}\) dust (B. Herut, personal communication, 2013).

\[
\text{Atmospheric Flux} = 31 \text{ g m}^{-2} \text{ yr}^{-1} \times 25 \mu \text{mol Si g}^{-1} = 775 \mu \text{mol m}^{-2} \text{ yr}^{-1}
\]

Using a total area of 1337 000 km\(^2\) this converts to \(1.0 \times 10^9\) mol Si yr\(^{-1}\).

The atmospheric flux was thus a relatively small direct source of silica to the EMS, though with potentially large variability since it is known that Saharan dust is very inhomogeneous. Some sources of desert dust are known to have higher and potentially much higher fractions of opaline silica. In particular the Bodele depression – which is the single largest source of Saharan dust at present, particularly to the...
Table 4. Total silica budget for the eastern Mediterranean Sea. The fluxes are in units of $10^9$ mol Si yr\(^{-1}\).

<table>
<thead>
<tr>
<th>Source of silica</th>
<th>Flux</th>
<th>Comment</th>
</tr>
</thead>
<tbody>
<tr>
<td>Straits of Sicily – dissolved silicic acid</td>
<td>121</td>
<td>Calculated using winter (Nov 1999) SA and water flow data for 6 months and summer (Sept 1997) silicic acid and flow data for 6 months (Table 3).</td>
</tr>
<tr>
<td>Straits of Sicily – BSi</td>
<td>16</td>
<td>Using summer flow rate ($&lt; 100$ m) and 1 µ mol L(^{-1}) during summer 6 months (Table 3).</td>
</tr>
<tr>
<td>Riverine input – dissolved silicic acid and BSi</td>
<td>27</td>
<td>Ludwig et al. (2009) corrected using global values for BSi input from Laruelle et al. (2009).</td>
</tr>
<tr>
<td>Submarine groundwater flux</td>
<td>9.7</td>
<td>Using estimates for freshwater flow into the EMS and silicic acid values from Laruelle et al. (2009) and Weinstein et al. (2011).</td>
</tr>
<tr>
<td>Sediment porewater flux</td>
<td>57</td>
<td>Calculated value for silicic acid flux assuming all is the result of diagenetic dissolution from aluminosilicate minerals.</td>
</tr>
<tr>
<td>Atmospheric Saharan dust input – BSi</td>
<td>1</td>
<td>Based on total BSi measured in Saharan dust samples (Krom, unpubl. data).</td>
</tr>
<tr>
<td><strong>Total input</strong></td>
<td><strong>232</strong></td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Source of silica</th>
<th>Flux</th>
<th>Comment</th>
</tr>
</thead>
<tbody>
<tr>
<td>Straits of Sicily – silicic acid</td>
<td>264</td>
<td>Measured average silicic acid $&gt; 100$ m in summer and $&gt; 200$ m in winter. Astraldi et al. (1999) for flow data and MTP-MATER data for silicate (Table 3).</td>
</tr>
<tr>
<td>Sediment burial flux</td>
<td>0</td>
<td></td>
</tr>
<tr>
<td><strong>Total outflow</strong></td>
<td><strong>264</strong></td>
<td></td>
</tr>
<tr>
<td>Net export flux from the eastern Mediterranean basin</td>
<td><strong>32</strong></td>
<td></td>
</tr>
</tbody>
</table>

eastern Atlantic –is a diatomite and contains relatively large amounts of opaline silica.

4.3 Total silica budget for the EMS

The net flux of dissolved silicic acid flowing out of the EMS at the Straits of Sicily is $(121 \times 10^9$ mol Si yr\(^{-1}\)), is somewhat lower than the lower range of the previous estimate of $157 \times 10^9$ mol Si yr\(^{-1}\) by Ribera d’Alcalà et al. (2003) because it includes BSi and winter upwelled silicic acid flowing into the EMS. Here detailed estimates have been made of the silica input from riverine input containing BSi, submarine groundwater flux, sediment porewater flux and atmospheric input. By including these terms, the deficit of silica in much reduced to $32 \times 10^9$ mol Si yr\(^{-1}\).

Key to this total budget is the assumption that the dissolution of externally supplied BSi, and the internal weathering of aluminosilicates are important processes in the EMS. Such dissolution is required to explain the initial concentration of dissolved silicic acid in the western Ionian basin, the increase in dissolved silicic acid across the ADW to the southeastern Levantine basin as well as to “close” the total silica budget for the basin. While such dissolution processes are known to occur elsewhere in the ocean (DeMaster, 2004), they are particularly important in the EMS with its low initial concentrations of silicic acid, relatively high temperature of deep water and high input of dust and riverine particles.

By contrast, recycling by diatoms appears not to be regionally important. The observed vertical profile in dissolved silicic acid can be explained entirely as a result of the circulation of the EMS, with the surface concentration of silicic acid being derived from surface water advecting in through the Straits of Sicily and the increase in deep water being due to processes during ADW formation in the S Adriatic and the flux of silicic acid from sediments across the basin. That is not to say that there is no internal cycling of silica from diatoms in the EMS. Crombet et al. (2011) showed clearly that there is a measured increase in diatom abundance and a “deep glass forest” where there is nutrient upwelling (i.e. in the Straits of Sicily, the Ionian front and possibly also the Cyprus eddy). There are also diatoms in the permanent cold-water eddies and increased diatom abundance in coastal waters. What is argued here is that the presence of diatoms and recycling from diatoms falling from the photic zone are not required to explain the regional pattern of silicic acid in the offshore EMS.

The remaining deficit in the EMS budget could possibly be explained simply as a result of the errors and assumptions which were made in the principal terms used in constructing
this silica budget. For example if the BSi in summer were an average of 2.0 µmole Si kg$^{-1}$ instead of the assumed value of 1 µmole Si kg$^{-1}$, the deficit would be eliminated; while if the average concentration of the outflowing water were that measured in September and October (6.2 µmol Si kg$^{-1}$) rather than the 7.4 µmol Si kg$^{-1}$ determined in November, there would be a small net accumulation of silica in the EMS. One potentially important process which has not been included in this budget is the effect of silica cycling processes in coastal areas. Recently it was shown that Si is advected from the shore towards the open sea in the eastern-most part of the Mediterranean (Efrati et al., 2013). If silicic acid from diagenesis on the continental shelf is advected from the coastal shelf into the offshore waters, it will represent a new source of silicic acid which will be exported from the basin. A particularly important source for such silicic acid may be the Nile cone and adjacent coastal shelves since it is known that, during the Nile flood (before the closure of the Aswan Dam), there was a major diatom bloom in the offshore region (Halim, 1991). This process could take place either by direct advection of silicic acid into intermediate water or by the export of BSi produced by diatom uptake of silicic acid in surface waters of the shelf being transported offshore and then dropping into the intermediate waters. We recognise that Table 4 represents a preliminary budget and that further targeted measurements need to be made of the major terms in this budget, particularly the annual BSi content in the surface inflowing waters at the Straits of Sicily, the sediment and diagenetic processes in the surficial sediments and riverine inputs.

Acknowledgements. Sandra Lynn Jones measured and evaluated interstitial silica concentrations in the eastern Mediterranean for her graduate thesis. Unfortunately she passed away before the scientific research context was developed that showed how significant her work really was. She was a fine colleague who had a remarkable creativity and will continue to be missed.

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